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o-(Hydroxyalkyl)phenyl P-Chirogenic Phosphines as Functional Chiral Lewis Bases

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ABSTRACT

The stereoselective synthesis of P-chirogenic phosphines bearing an o-hydroxyalkyl chelating arm is described. The synthesis is based either on the hydroxyalkylation of P-chirogenic o-bromophenylphosphines (borane) or on their carbonatation and then reduction. The hydroxyalkylation with benzaldehyde or pivalaldehyde affords a mixture of epimers which are isolated by chromatography and characterized by their X-ray structures. Preliminary assays of free P-chirogenic o-(hydroxyalkyl)phenyl phosphines, as new functional Lewis bases in catalyzed asymmetric aza-MBH reaction, lead to β -aminoester derivatives with ee values up to 74%.

A chiral organophosphorus bearing a heteroatom or a functional group in the *ortho* position of an aryl substituent is a useful ligand or Lewis base for the development of catalyzed or organocatalyzed asymmetric reactions. 1-4

Among the chiral functional organophosphorus derivatives, those bearing a phenol substituent have gained attention because they could be used in various asymmetric reactions using either transition-metal catalysts³ or bifunctional organocatalysts.⁴ Despite the importance of the functional chiral organophosphorus compounds, few efficient stereoselective syntheses have been described to date. The synthesis of a P-chirogenic organophosphorus bearing a hydroxyalkyl (or aryl) substituent could be achieved by demethylation of a methoxy substituent, ^{2c,3a} by carbonatation then reduction,⁵ or by trapping a C,Ophenol dilithium reagent with a chlorophosphine.⁶ A more convenient method has been developed on the basis of the ortho Fries-like rearrangement of the corresponding 2-bromoaryl phosphinite borane, with retention of configuration at the phosphorus atom.⁷

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Scheme 1. Stereoselective Synthesis of P-Chirogenic o-Bromophenyl Phosphines Using Aryne Chemistry

We have recently described an efficient stereoselective synthesis of *o*-bromophenyl P-chirogenic phosphines **5**, based on the reaction between the secondary phosphine borane **1** and benzyne **2**, generated *in situ* from 1,2-dibromobenzene **3** by metal—halide exchange (Scheme 1).⁸ In continuity of this research, we were interested in the synthesis of P-chirogenic phosphines bearing a hydroxyalkyl chelating arm, for the development of a new class of chiral *o*-functional Lewis bases.

Herein, we report an efficient stereoselective synthesis of P-chirogenic *o*-(hydroxyalkyl)phenyl phosphines, either by hydroxyalkylation of the corresponding *o*-bromophenyl phosphine precursors **5** or by Fries-like rearrangement of a phosphinite borane derived from 2-bromobenzyl alcohol.

In the first case, the hydroxyalkylation was achieved by the reaction of the anion derived from the *o*-bromo phenylphosphine borane **5a**, or the free phosphine **5b**, with the benzaldehyde **6a** or the pivalaldehyde **6b**. The corresponding P-chirogenic *o*-(hydroxyalkyl)phenyl phosphines (borane) were obtained as a mixture of two epimers **7** and **7**′ with respect to the absolute configuration at the P-center, with yields ranging from 42% to 72% (Table 1).

Thus, the reaction of the phosphine borane (R_p) -5a with the benzaldehyde 6a affords the epimeric mixture (S_p) -7a and (S_p) -7a' in the ratio 45:55 (entry 1). The epimers were separated by chromatography, and their enantiomeric purities were checked by HPLC on a chiral column. Crystals of (S_p) -7a' were grown from methylene chloride/hexane as solvent, and its drawing is shown in Figure 1. The structure of (S_p) -7a' shows a distorted tetrahedral geometry at the P-atom which is typical of phosphine borane adducts, which proves the (S_p) - and (R)-configuration at the P- and C-atom, respectively. These absolute configurations are supported by refinement of the Flack parameter (Table S1).

In the case of the reaction of the o-bromo ferrocenylphenylphosphine borane (S_p) -5a with benzaldehyde 6a, the o-(hydroxyalkyl)phenylphosphine borane (R_p) -7a and

Table 1. Synthesis of P-Chirogenic o-(Hydroxyalkyl)phenyl Phosphines (Borane) $7\mathbf{a} - \mathbf{d}^a$

entry	${ m reagents}^a$	ι	R^1	\mathbb{R}^2	\mathbb{R}^3	Y	$products^d$	ratio ^e 7:7 ′	•
1	$(R_{\rm p})$ -5 ${f a}^{a,g}$ (6a	Fc	Ph	Ph	BH_3	$(S_{\rm p})$ -7a, 7a'	45:55	71
2	$(S_{\rm p})$ -5 ${\bf a}^{a,h}$ 6	6a	Ph	Fc	Ph	BH_3	$(R_{\rm p})$ -7a, 7a'	64:36	72
3	$(S_{\rm p})$ -5 ${\bf a}^{a,h}$ 6	6b	Ph	Fc	$t ext{-Bu}$	BH_3	$(R_{\rm p})$ -7b, 7b'	60:40	66
4	$(S_{\rm p})$ -5 ${\bf b}^{a,h}$ (6a	Ph	o-An	Ph		$(R_{\rm p})$ -7c, 7c'	$80{:}20^i$	45
							$(R_{\rm p})$ -7c, 7c'	$70:30^{i}$	48
	$(S_{\rm p})$ -5 ${\bf b}^{c,h}$ 6						$(R_{ m p})$ -7c, 7c $'$	$70:30^{i}$	42
7	$(S_{\rm p})$ -5 ${\bf b}^{a,h}$ (6b	Ph	o-An	t-Bu		$(R_{\rm p})$ -7d, 7d'	$60:40^{i}$	50

^a Reaction conditions: **5** (0.4 mmol), *n*-BuLi (0.44 mmol) in THF (2 mL) at −78 °C for 1 h. Aldehyde (0.8 mmol for benzaldehyde **6a** or 1.6 mmol for pivalaldehyde **6b**) in dry THF (0.5 mL) was added at −78 °C, and the mixture was warmed to rt and stirred again for 1 h 30 min. ^b Benzaldehyde **6a** was added at 0 °C. ^c Toluene solution of benzaldehyde **6a** was added. ^d ee >99% determined by HPLC on chiral column. ^e Determined after isolation by chromatography. ^f Isolated yields after purification by column chromatography. ^g Prepared from (−)-ephedrine. ^h Prepared from (+)-ephedrine. ⁱ The relative configuration was attributed by comparison with the major epimers (R_p S)-7**a** and (R_p S)-7**b**.

 $(R_{\rm p})$ -7a' were obtained in a 64:36 ratio, 72% yield and with 99% enantiomeric excess (entry 2). The structure of the major isomer has also been determined by single crystal X-ray diffraction, as the enantiomer $(R_{\rm p},S)$ -7a of the o-(hydroxyalkyl)phenyl phosphine borane $(S_{\rm p},R)$ -7a' shows in Figure 1.

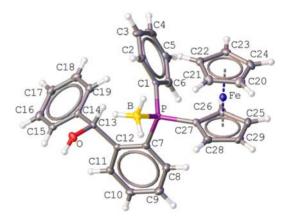


Figure 1. OLEX2 view of the *o*-(hydroxymethyl)phenyl phosphine (S_p,R) -7a', showing thermal ellipsoids at the 50% probability level. Selected bond lengths [Å], angles [deg]: C1–P 1.818(4); C7–P 1.831(4); C13–O 1.429(4); C27–P 1.792(4); P–B 1.913(4); C2–C1–P 120.1(3); O–C13–C14 112.8(3); O–C13–C12 110.6(3); C27–P–C1 105.26(16); C27–P–C7 105.91(17); C1–P–C7 104.03(16); C27–P–B 110.35(18); C7–P–B 113.95(18). B–P–C27–C28 30.90(4); B–P–C1–C2 3.5(4); B–P–C7–C12 63.5(3); C7–C12–C13–O –101.4(4).

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When the reaction was achieved with the ferrocenyl-(o-bromophenyl)phosphine borane (S_p) - $\mathbf{5a}$ and the pivalaldehyde $\mathbf{6b}$, the corresponding o-(hydroxyalkyl) phenyl phosphine borane (R_p) - $\mathbf{7b}$ (and $\mathbf{7b}'$) was obtained in 66% yield and with a 60:40 ratio (entry 3). The structure of the major isomer (R_p) - $\mathbf{7b}$ has also been determined by single crystal X-ray diffraction (Figure 2).

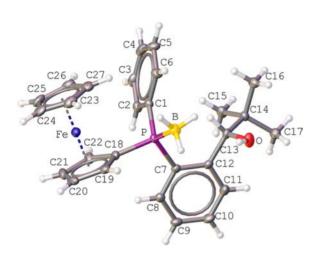


Figure 2. OLEX2 view of the *o*-(hydroxyalkyl)phenylphosphine (R_p ,S)-**7b**, showing thermal ellipsoids at the 50% probability level. Selected bond lengths [Å], angles [deg]: C1-P 1.818(4); C7-P 1.842(4); C13-O 1.436(5); C18-P 1.792(4); P-B 1.934(5); C2-C1-P 121.2(3); O-C13-C14 108.5(3); O-C13-C12 109.7(3); C18-P-C1 106.5(2); C18-P-C7 104.9(2); C1-P-C7 106.3(2); C18-P-B 110.2(2); C7-P-B 112.9(2); B-P-C18-C22 163.4(4); B-P-C1-C2 173.0(4); B-P-C7-C12-56.3(4); C7-C12-C13-O 108.0(4).

This structure shows again a distorted tetrahedral geometry at the P-atom which is typical of phosphine borane adducts, proving the (*R*)- and (*S*)-configuration at the P- and C-atom, respectively. These absolute configurations are supported by refinement of the Flack parameter (Table S1).

The structure of the minor isomer (R_p) -7b' was also determined by single crystal X-ray diffraction, and its drawing is shown in Figure 3. In this case, the X-ray structure proves the (R)-configuration at the P- and C-atom, which is supported by refinement of the Flack parameter (Figure 3, Table S1). After decomplexation of the borane complex using DABCO at room temperature, the anion obtained after metal-halide exchange of the o-anisyl-o-bromophenyl phenyl phosphine (S_p) -5b reacts with the benzaldehyde 6a to afford the corresponding o-(hydroxylalkyl)phenyl derivatives 7c, 7c' in 45% yield (entry 4). When the reaction was performed at rt, or in a 4:1 toluene/THF mixture, 7c and 7c' were obtained in a 70:30 ratio, showing the poor influence of the temperature or solvent on the diastereoselectivity (entries 5, 6). In the case of the reaction of (S_p) -5b, with the pivalaldehyde 6b, the corresponding o-(hydroxylalkyl)phenyl derivatives 7d and 7d' were obtained in 50% yield and with a 60:40 ratio (entry 7).

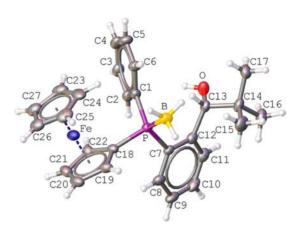


Figure 3. OLEX2 view of *o*-(hydroxymethyl)phenylphosphine borane (R_pR)-7b' showing thermal ellipsoids at the 50% probability level. Selected bond lengths [Å], angles [deg]: C1-P 1.816(4); C7-P 1.844(4); C13-O 1.440(5); C18-P 1.791(4); P-B 1.923(5); C2-C1-P 121.5(3); O-C13-C14 106.9(4); O-C13-C12 110.7(4); C18-P-C1 104.4(2); C18-P-C7 103.5(2); C1-P-C7 103.8(2); C18-P-B 111.4(2); C7-P-B 118.3(2); B-P-C18-C22 157.4(4); B-P-C1-C2 166.4(4); B-P-C7-C12-77.5(4); C7-C12-C13-O-124.7(4).

On the other hand, as the reaction of paraformaldehyde was sluggish, the stereoselective synthesis of P-chirogenic ferrocenyl-o-(hydroxymethyl) phenyl phosphine borane (S_p) -7e was achieved in 40% overall yield, by trapping the anion derived from the o-bromophenyl phosphine borane (R_p) -5a with CO_2 , followed by the reduction of the carboxylic acid derivative (S_p) -8 with $BH_3 \cdot DMS$ (Scheme 2a). The enantiomeric purity of the compound (S_p) -7e was checked on a chiral column in regards to a racemic sample, and no racemization was found to occur during this process.

Interestingly, the stereoselective synthesis of the ferrocenyl-o-(hydroxymethyl)phenylphosphine (S_p)-7e could also be achieved by the intramolecular *ortho* Fries-like rearrangement of the 2-bromobenzylphosphinite borane (R_p)-11, mediated by basic conditions (Scheme 2b).

The phosphinite borane **11** was prepared in 51% yield by reaction of the o-bromobenzyl alcohol **10** with the P-chirogenic chlorophosphine (S_p) -**9**, previously obtained using the (+)-ephedrine methodology. Finally, the phosphine borane derivative (S_p) -**7e** was obtained in 58% yield by rearrangement mediated by tert-butyllithium (Scheme 2b). The stereochemistry with retention of the configuration at the P-atom in the Fries-like rearrangement which was previously described in literature allows the (S)-configuration to be attributed to **7e**.

During the past decade, chiral phosphines bearing protic moieties have emerged as powerful catalysts for various asymmetric reactions with C–C bonds formation [Michael, Baylis–Hillmann, aza-Morita–Baylis–Hillmann (aza-M-B-H)]. To the best of our knowledge, the use of

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Scheme 2. Stereoselective Synthesis of the P-Chirogenic Ferrocenyl-o-(hydroxymethyl)phenylphosphine (S_p)-7e

(a)
$$F_{C}$$
 F_{C} F_{C}

P-chirogenic phosphines as an organocatalyst has been minimally considered to date in this area. In this context, the P-chirogenic o-(hydroxyalkyl)phenylphosphines (S_p ,S)-14 and (R_p ,R)-7c' were preliminarily evaluated as a functional Lewis base in the asymmetric aza-MBH reaction (Scheme 3).

The free phosphine (S_p,S) -14 was prepared in 98% yield by decomplexation of (S_p,S) -7a with DABCO. When the acrylate 12 reacts with the *N*-tosyl aldimine 13 in TBME at 5 °C, in the presence of 5 mol % of (S_p,S) -14, the unsaturated β -aminoester (S)-15 was obtained in 90% yield and with 74% ee (Scheme 3). In the case of the use of (R_p,R) -7c' as an organocatalyst, the β -aminoester (S)-15 was obtained in 82% yield and with 68% ee.

In conclusion, we have developed an efficient stereoselective synthesis of P-chirogenic o-(hydroxyalkyl) phenyl phosphine boranes. This synthesis is based on the hydroxyalkylation of the anion formed by metal—halide

Scheme 3. Asymmetric aza-MBH Reaction Catalyzed by the P-Chirogenic o-(Hydroxyalkyl)phenyl Phosphine (S_p,S) -14

$$β$$
-NpO + H $\frac{(S_p, S)$ -14 (5 mol %)}{5 °C, TBME, 60 h} β-NpO NHTs $\frac{(S_p, S)$ -15, 90%, ee = 74%

exchange from the P-chirogenic o-bromophenyl phosphines (free or as borane complexes). Under these conditions, a mixture of epimers was obtained in a 45:55 to 80:20 ratio, and they were separated by chromatography. The absolute configurations were determined by the X-ray structure of the P-chirogenic phosphine borane complexes. The stereoselective synthesis of P-chirogenic ferrocenylo-(hydroxymethyl) phenyl phosphine borane could be achieved either from o-bromophenyl phosphine borane by reaction with CO₂ and then reduction with borane or from 2-bromobenzyl phosphinite borane, by an intramolecular Fries-like rearrangement. The use of the P-chirogenic o-(hydroxyalkyl)phenyl phosphines as new P-chirogenic functional Lewis bases for an asymmetric aza-MBH reaction leads to the unsaturated β -aminoester 15 with ee values up to 74% in preliminary assays, prompting further development in this area soon.

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Supporting Information Available. Experimental procedures and spectral data for all new compounds. This material is available free of charge via the Internet at http://pubs.acs.org.

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The authors declare no competing financial interest.